Rhodium-Catalysed Enantioselective Synthesis of 4-Arylchroman-2-ones

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The rhodium-catalysed enantioselective 1,4-addition of organoboron reagents to arylidene Meldrum’s acids as acceptors, allows convenient access to 4-arylcroman-2-ones with good to excellent levels of enantioselectivity. The use of silyl-protected dioxaborinanes as donors was found to be advantageous to achieving good yields of product under anhydrous conditions.

The enantioselective construction of C-C bonds using the rhodium-catalysed 1,4-addition of organometallics is established as an important tool for organic synthesis. For the addition of aryl and alkénylboronic acids, the reaction is routinely carried out in aqueous solvents and can afford excellent enantioselectivities across a wide range of alkene acceptors. In the majority of applications, boronic acids are the coupling partners of choice for conjugate addition reactions. However, there can be issues with purification and manipulation. Often, an excess of reagent has to be used due to competing protodeboronation processes and by competing formation of trimeric cyclic anhydrides (boroxines) in solution, leading to difficulties in being able to accurately measure reaction stoichiometry. A number of elegant solutions to this problem have been presented involving the use of preformed boronate reagents that can be isolated and stored prior to use including tris(hydroxy)borates, lithium trimethoxyboronate species, trifluoroborate salts (such as 4) and N-methyliminodiacetic acid (MIDA) boronates. An important addition to this range of donors are the cyclic triolborates synthesized by Miyaura. These boronate reagents are reported to be stable in air and water and more soluble in organic solvents than potassium trifluoroborates. In this paper we describe the utility of silyl-protected dioxaborinanes in rhodium-catalysed 1,4-addition reactions under anhydrous conditions. This allows the enantioselective addition to arylidene Meldrum’s acid derivatives and a subsequent asymmetric synthesis of 4-arylcroman-2-ones. Fillion and co-workers have shown that aryldiene Meldrum’s acid derivatives such as 1a are useful substrates for the rhodium-catalysed 1,4-addition of organozinc donors and organotin donors generating a range of products with significant scope for further diversification. The lack of literature reports for the addition of organoboron reagents reflects a greater challenge due to the sterically hindered nature of the trisubstituted alkene and the sensitivity of the malonate functionality in the product to attack by nucleophiles (e.g. water). Indeed, our initial investigation into the rhodium-catalysed addition of phenylboronic acid 3 to the 4-methoxymethyl arylidene Meldrum’s acid derivative 1a in dioxane at room temperature afforded low conversion to product 2. The anhydrous conditions and low temperature proved detrimental to the application of potassium trifluoroborate salt 4 and hydroxymethyl dioxaborinane 6 (Scheme 1).
acceors that possessed electron-donating ether substituents (Scheme 2, 9-16).

Scheme 2. Scope of addition using silyl-protected dioxaborinanes.

In contrast, aryldiene Meldrum’s acid derivative 8 with the para-F group on the aryl ring was unreactive under the standard reaction conditions. Similarly, other electron-withdrawing substituents (Br and NO₂) were not tolerated on the acceptor. This observation can be rationalised by the inductive deactivation of the hindered trisubstituted alkene derivative to carbometallation. Substitution of the silyl-protected dioxaborinane was possible at the ortho, meta, and para positions. However, yields were compromised when both donor and acceptor were ortho-substituted (Scheme 2, 11-13). With the optimised set of reaction conditions, we next established that silyl-protected dioxaborinane 7 could be employed for the addition to typical cyclic activated alkenes (Scheme 3).

Scheme 3. The addition to cyclic activated alkenes.

The new organoboron reagent proved to be remarkably effective for the rhodium-catalysed 1,4-addition to cyclic substrates under anhydrous conditions at room temperature. The stoichiometry of the organoboron donor could be reduced to just 1.1 equivalents and the products (Scheme 3 17-20) were obtained in high isolated yield. The 4-arylchroman-2-one system is of synthetic interest as it features in a number of natural flavonoid structures or as intermediates in drug synthesis. The rhodium-catalysed 1,4-addition of boronic acids to coumarin derivatives has been demonstrated to afford 4-arylchroman-2-ones with high enantioselectivity. However, coumarins tend to be relative poor substrates for conjugate additions, thus up to ten equivalents of boronic acid donor were required for successful conversion to product. Our alternative strategy shown in Figure 1 is based on an enantioselective rhodium-catalysed 1,4-addition to (2-benzyloxy)phenyl aryldiene Meldrum’s acid derivative followed by hydrolysis/decarboxylation of the cyclic malonate, deprotection of the benzyl ether and intramolecular esterification to provide the desired 4-arylchroman-2-one products.

Figure 1. Enantioselective synthesis of 4-arylchroman-2-ones.

Scheme 4. Enantioselective additions with (R,R)-Ph-bod*.

The development of an enantioselective rhodium-catalysed 1,4-addition to (2-benzyloxy)phenyl alkylidene Meldrum’s acid derivative proved to be challenging. Preliminary attempts using a [Rh(C₂H₅)₂Cl₂]₂ precatalyst in the presence of atropisomeric...
bidentate phospine ligands afforded no desired product, both the silyl-protected dioxaborinane donor and the alkene were recovered intact.

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\text{(i) DMP, 74% HCl (10:1), 100 °C; (ii) 10% Pd(OAc)\textsubscript{2}, H\textsubscript{2} (1 atm), EtOAc; (iii) pTSA, benzene, 80 °C.}
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**Scheme 5.** Enantioselective synthesis of 4-arylchroman-2-ones.

**Figure 2.** ORTEP drawing of (R)-4-(4-chlorophenyl)chroman-2-one 20.

Enantiopure diene ligands can substitute for chiral phosphines in enantioselective processes and often show superior reactivity and enantioselectivity.\(^{14}\) Fortunately, the enantiopure diene (R,R)-Ph-bod* introduced by Hayashi and co-workers allowed the arylation to proceed with good reactivity and enantioselectivity under anhydrous conditions (Scheme 4).\(^{15}\) The isolated yields obtained with the enantiopure diene ligand (R,R)-Ph-bod* were consistent with the racemic protocol. Substitution of the silyl-protected dioxaborinane at the para position afforded excellent yields of product (Scheme 4, 11, 14 and 15) with lower yields for the meta substituted aryl group in 12 and poor yields for the ortho,ortho substituted product 13. The enantioselectivity was determined after conversion to the 4-arylchroman-2-one products 16-21. Thus, the enantoenriched Meldrum’s acid derivative 10-15 was heated in a stirring mixture of DMF and aqueous HCl to afford the carboxylic acid. Hydrogenation of the crude reaction mixture in ethyl acetate in the presence of palladium impregnated on carbon led to deprotection of the benzyl ether. The resulting phenol was then heated in the presence of a catalytic amount of para-toluenesulfonic acid to afford the desired product 16-21 over three steps (Scheme 5). The enantioselectivity was good for a range of silyl-protected dioxaborinane donors (up to 97% ee for 20). The enantioselectivity was significantly lower when both donor and acceptor were ortho-substituted (Scheme 5, 9). The absolute configuration of 20 was determined to be (R)- by X-ray crystallography (Figure 2). A recent density functional theory study of the rhodium-catalysed addition of phenylboronic acid to cyclohexenone offers useful insight into the origin of enantioselectivity with Rh(1)/Ph-bod* complexes.\(^{16}\)

**Conclusions**

In conclusion, we have shown that silyl-protected dioxaborinanes perform exceptionally well as donors in rhodium-catalysed 1,4-addition reactions under anhydrous conditions. In the scenario presented here, this allowed an enantioselective addition to aryldiene Meldrum’s acid derivatives and a subsequent asymmetric synthesis of 4-arylchroman-2-ones. Further studies to explore the application of these new donors in additions to other challenging alkenic acceptors and to explain their unique properties is in progress.

**Notes and references**

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[2] Electronic Supplementary Information (ESI) available: Experimental procedures, characterisation data, copies of NMR spectra for compounds synthesised in this study can be found in the Supporting Information. See DOI: 10.1039/b000000x/

[3] CCDC 836829 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

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